

AMENDMENTS TO THE CLAIMS:

This listing of claims will replace all prior versions, and listings, of claims in the application:

LISTING OF CLAIMS:

1. (Currently Amended) A dehydrofluorination process to convert an aromatic carbamoyl fluoride to the corresponding isocyanate, the process comprising ~~subjecting~~ gradually introducing carbamoyl fluoride ~~to a temperature at least equal to 80°C, in a dissolved or finely dispersed state in a solvent and at a~~ into a solvent heel ~~at a~~ temperature of at least 80°C, ~~wherein said carbamoyl fluoride is in the dissolved or finely dispersed state in the solvent.~~

2. (Currently Amended) The process as claimed in Claim 1, wherein the ~~said reaction~~ temperature of the solvent heel is at most equal to 150°C.

3. (Previously Presented) The process as claimed in Claim 1, wherein said solvent exhibits a boiling point of at least 100°C.

4. (Previously Presented) The process as claimed in Claim 1, wherein the reaction is carried out at a pressure such that, at the reaction temperature, the solvent is boiling.

5. (Previously Presented) The process as claimed in Claim 1, wherein the solvent is miscible with hydrofluoric acid which does not react with the carbamoyl fluoride.

6. (Previously Presented) The process as claimed in Claim 1, wherein said carbamoyl fluoride is introduced into the solvent with hydrofluoric acid.

7. (Previously Presented) The process as claimed in Claim 6, wherein the ratio of the hydrofluoric acid to the carbamoyl fluoride (HF/carbamoyl fluoride) is at least equal to 2.

8. Canceled

9. (Previously Presented) The process as claimed in Claim 1, wherein the addition of the carbamoyl fluoride to the solvent is carried out at a rate such that, in the ten final 90% of the reaction duration taking place below 100°C, the molar ratio of hydrofluoric acid to isocyanate (HF acid/aromatic isocyanate) is always less than 0.5.

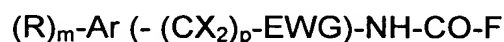
10. (Currently Amended) The process as claimed in Claim 1, wherein the carbamoyl fluoride substrate comprises an aliphatic carbon, that is sp^3 sp³ hybridized, bearing at least two fluorines.

11. (Previously Presented) The process as claimed in Claim 10, wherein said aliphatic carbon bearing at least two fluorines is a benzyl carbon and is directly attached to an aromatic ring.

12. (Previously Presented) The process as claimed in Claim 11, wherein said aromatic ring is that bearing the nitrogen of the carbamoyl functional group.

13. (Previously Presented) The process as claimed in Claim 1, wherein the reaction mixture comprises less than 1%, with respect to the starting carbamoyl fluoride, expressed as moles, of impurities exhibiting a chlorine in the benzyl position.

14. (Currently Amended) The process as claimed in Claim 1, wherein the carbamoyl fluoride corresponds to the formula:



where:

- Ar is an aromatic residue;
- the X units, which are alike or different, represent a fluorine or a radical of formula $\text{C}_n\text{F}_{2n+1}$ with n an integer at most equal to 5;
- p represents an integer at most equal to 2;
- EWG represents a hydrocarbonaceous group or an electron-withdrawing group, the optional functional groups of which are inert under the reaction conditions;

The the total carbon number of $\text{-(CX}_2\text{)}_p\text{-EWG}$ is between 1 and 15;

- m is 0 or an integer from 1 to 4;
- R represents alike or different radicals comprising halogens or hydrocarbonaceous radicals.

15. (Currently Amended) The process as claimed in Claim 1, wherein the ~~solvents are selected from the group consisting of chlorobenzenes~~ solvent is a chlorobenzene.

16. (Previously Presented) The process as claimed in Claim 15, wherein the chlorobenzene is a monochlorobenzene, a dichlorobenzene or a trichlorobenzene.

17. (Currently Amended) A dehydrofluorination process to convert an aromatic carbamoyl fluoride to the corresponding isocyanate, the process comprising subjecting carbamoyl fluoride to a temperature at least equal to 80°C, by gradually introducing the carbamoyl fluoride in a dissolved or finely disposed state in a solvent ~~and at a temperature of at least 80°C, wherein said carbamoyl fluoride is in the dissolved or finely dispersed state in the solvent, and~~ wherein the carbamoyl fluoride is introduced into the solvent with hydrofluoric acid in the form of a solution comprising anhydrous hydrofluoric acid.

18. (Currently Amended) A dehydrofluorination process to convert an aromatic carbamoyl fluoride to the corresponding isocyanate, the process comprising subjecting carbamoyl fluoride to a temperature at least equal to 80°C, by gradually

introducing the carbamoyl fluoride in a dissolved or finely disposed state in a solvent and at a temperature of at least 80°C, wherein said carbamoyl fluoride is in the dissolved or finely dispersed state in the solvent, and wherein so that the total yield of isocyanate is at least about 70%.

19. (New) A dehydrofluorination process to convert an aromatic carbamoyl fluoride to the corresponding isocyanate, the process comprising gradually introducing carbamoyl fluoride in a dissolved or finely dispersed state in a solvent into a solvent heel at a temperature of at least 80°C and wherein during introduction of the carbamoyl fluoride the ratio of hydrofluoric acid to carbamoyl fluoride is at least equal to 2.

20. (New) The process of Claim 19, wherein the ratio of hydrofluoric acid to carbamoyl fluoride is at least equal to 3.

21. (New) The process of Claim 19, wherein the ratio of hydrofluoric acid to carbamoyl fluoride is at least equal to 4.

22. (New) A dehydrofluorination process to convert an aromatic carbamoyl fluoride to the corresponding isocyanate, the process comprising gradually introducing carbamoyl fluoride in a dissolved or finely dispersed state in a solvent into a solvent heel at a temperature of at least 80°C and wherein the ratio of hydrofluoric acid, including free hydrofluoric acid and added hydrofluoric acid, to isocyanate groups, real or masked in carbamoyl fluoride form, is at most equal to 5.

23. (New) The process of Claim 22, wherein the ratio of hydrofluoric acid to isocyanate groups is at most equal to 0.3.

24. (New) The process of Claim 22, wherein the ratio of hydrofluoric acid to isocyanate groups is at most equal to 0.1.